





Dr. P. Jagadeesan Professor of Civil Engineering, GNITC No verified email Design of RC structures



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Non-Linear Pushover Analysis of RC Frame under Static Lateral Loading P JAGADEESAN, T PALANISAMY International journal of earth sciences and engineering 8 (4), 1899-1704		2015









## Strengthening of brick masonry using basalt fiber reinforced cement mortar

January 2015

#### Authors:





T. Palanisamy





Citations (2)

References (3)

### Abstract

Brick masonry is one of the primary structures and it plays a role in Reinforced Concrete (RC) frame structure. It is very weak in tension and has low ductility response. Normally in brick masonry, cement mortar reaches failure before brick attains the failure. For strengthening the brick masonry, it is essential to increase the strength of the cement mortar. Basalt fiber is added with cement mortar in different proportions as 0.5%, 1% and 1.5% of weight of cement. Compressive strength and Young's modulus of brick masonry were tested with and without of basalt fiber in cement mortar. Similarly, cube compressive strength of cement mortar was also tested in the same manner. The experimental results show that 1% of basalt fiber in the cement mortar gives the optimum value of properties of the brick masonry.

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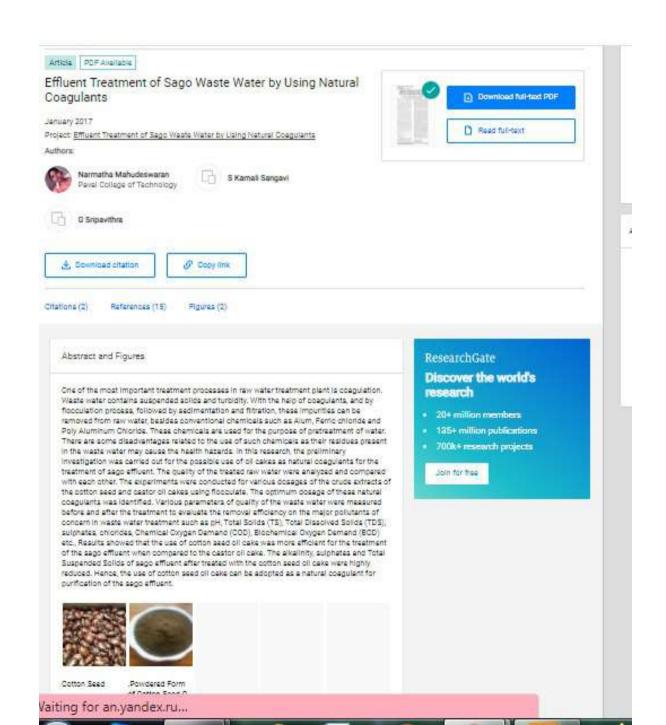
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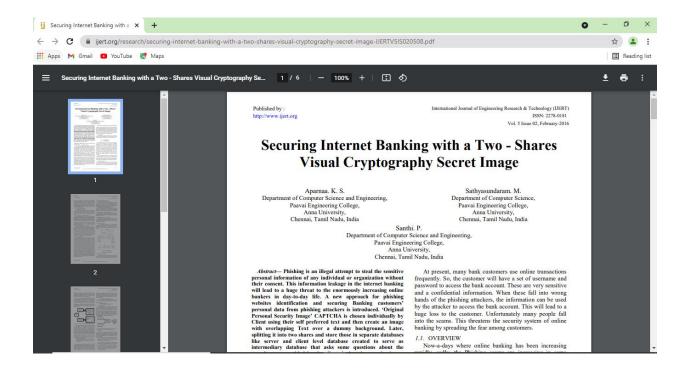
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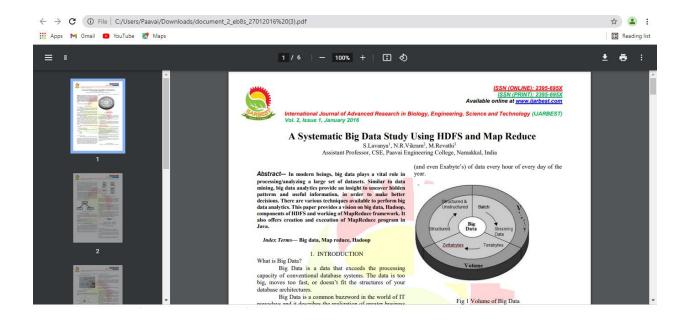
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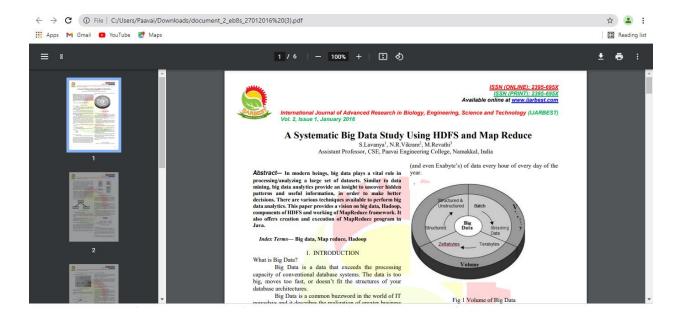




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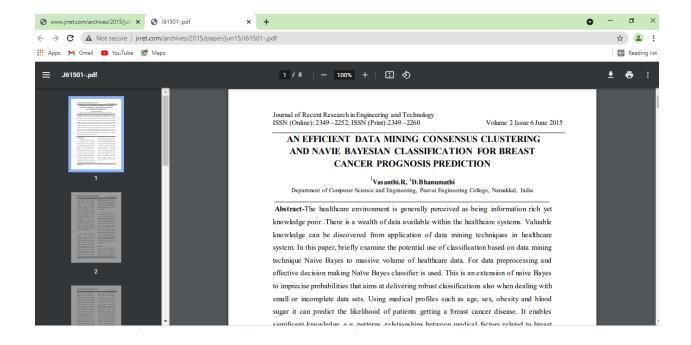


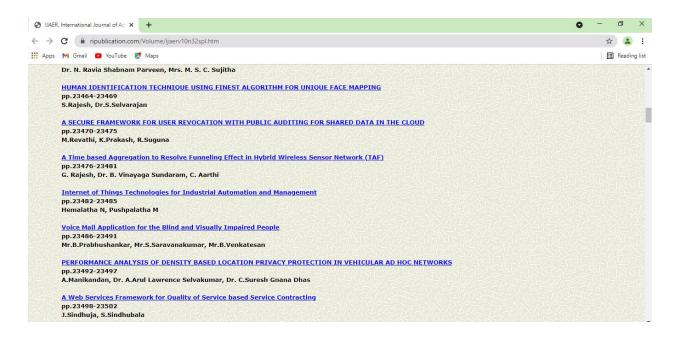


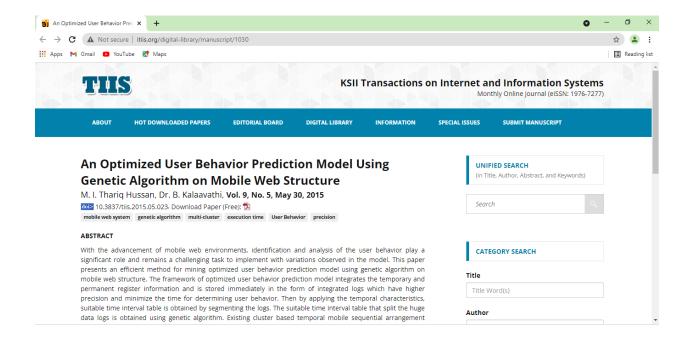


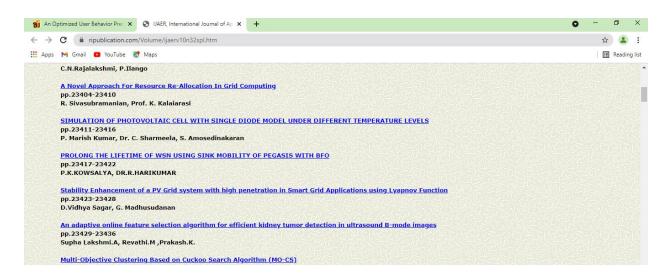


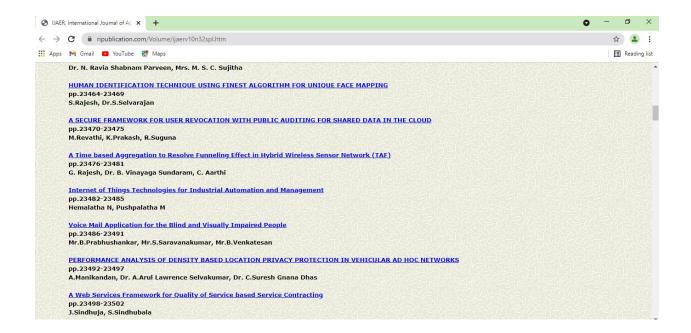






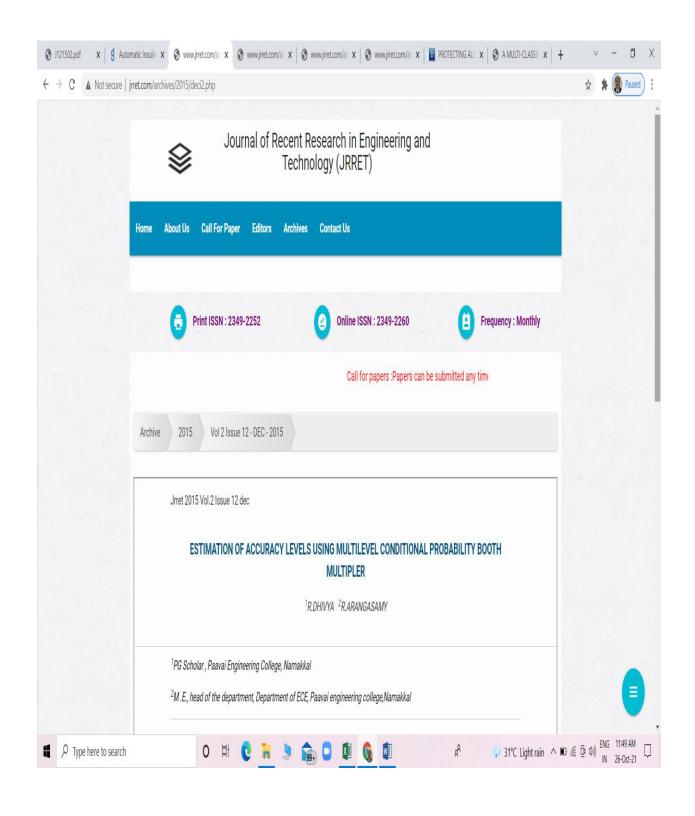


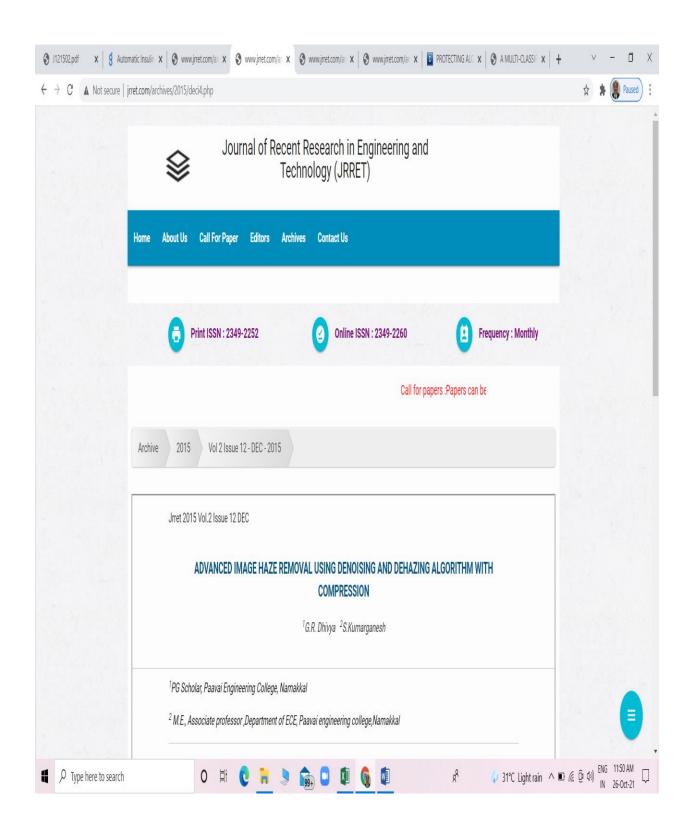


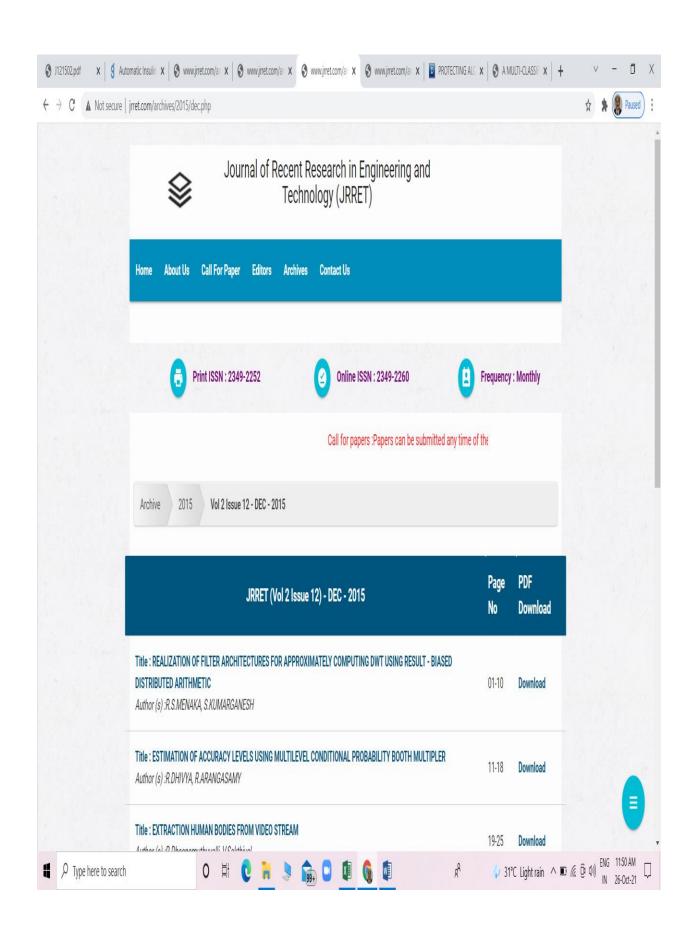


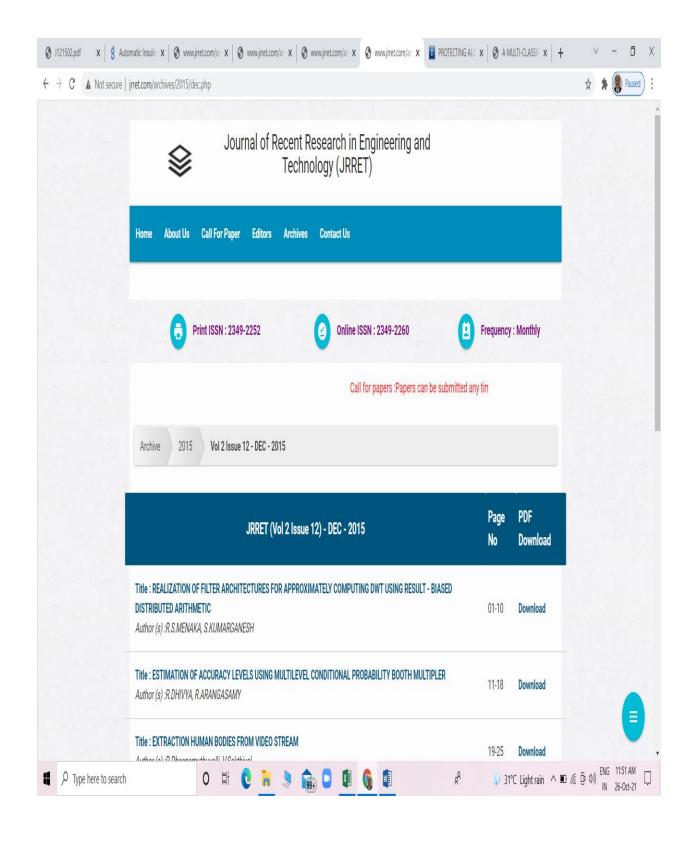


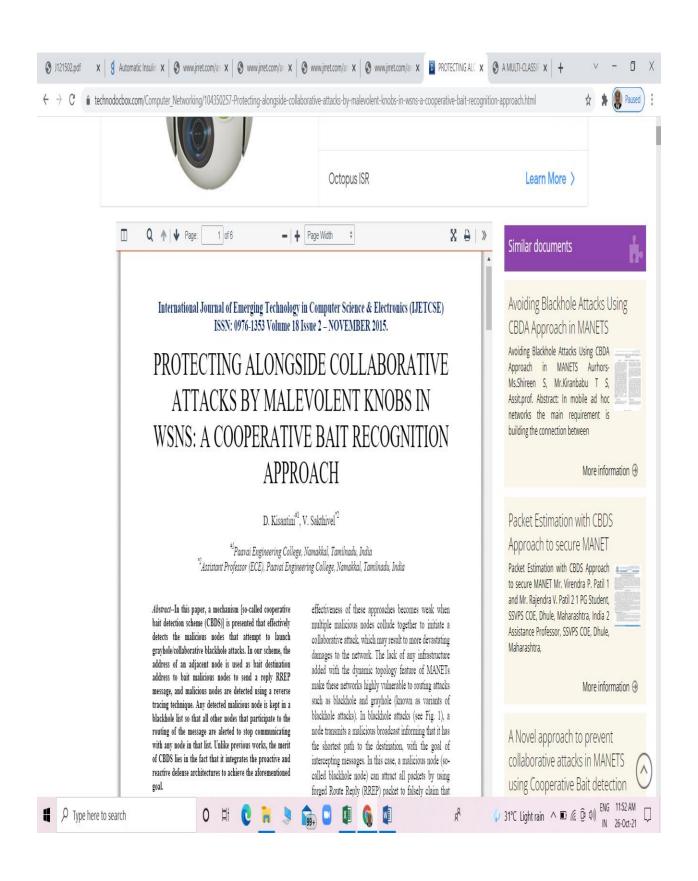


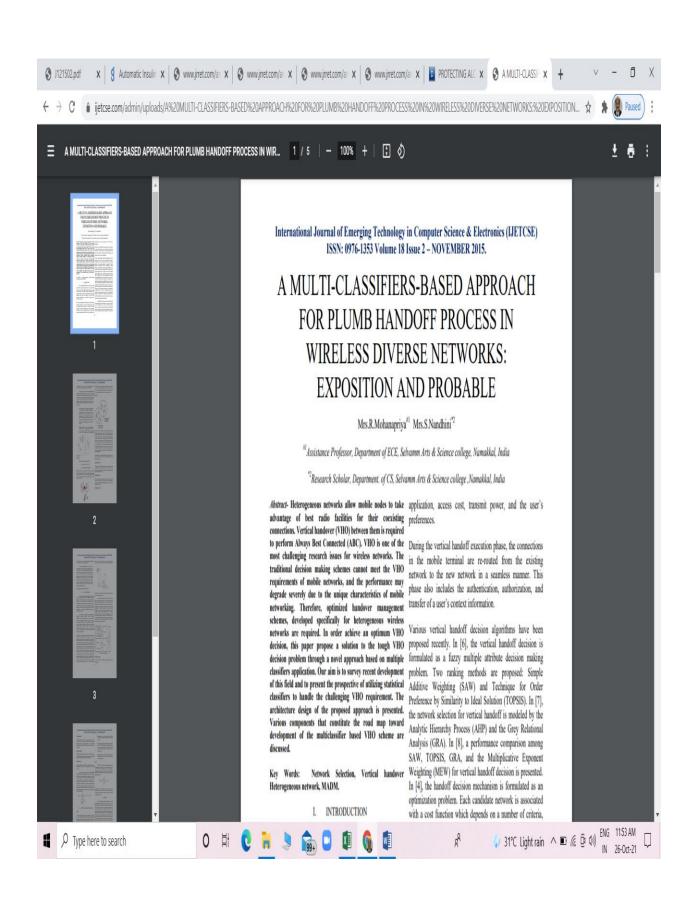


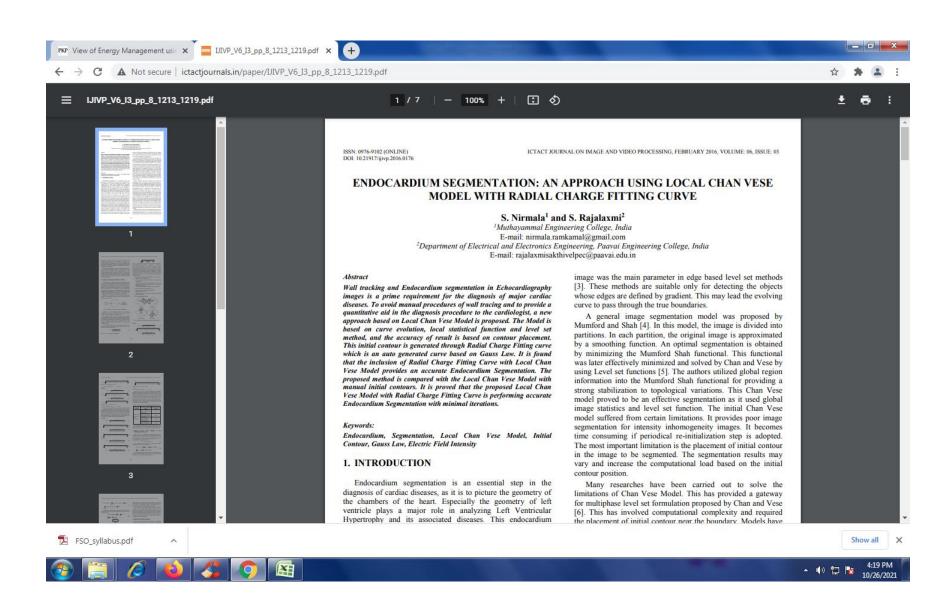




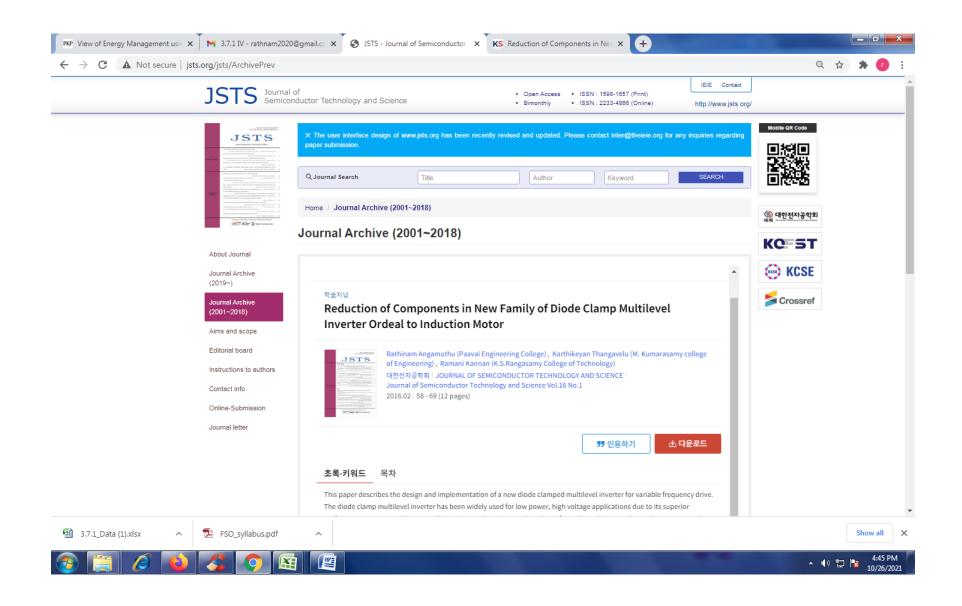


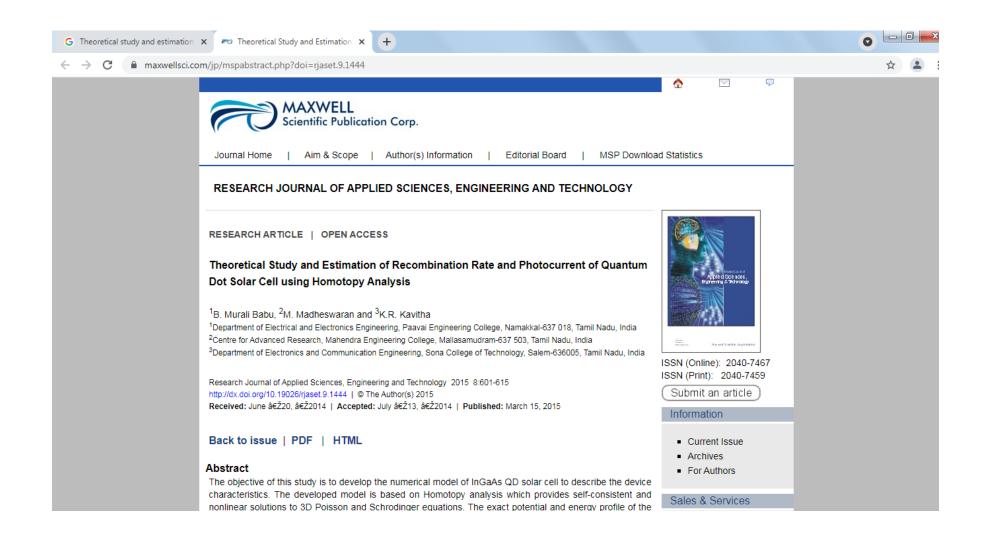


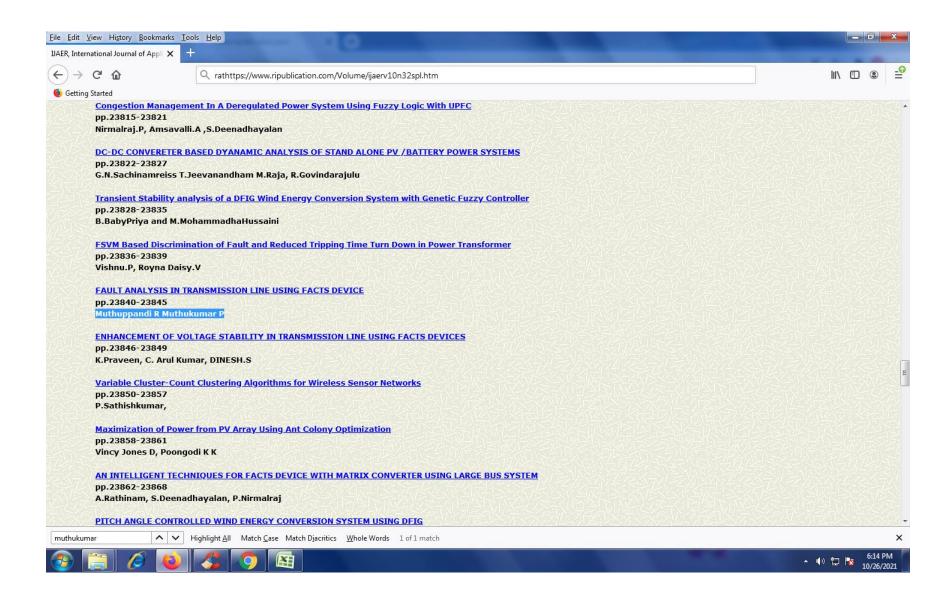


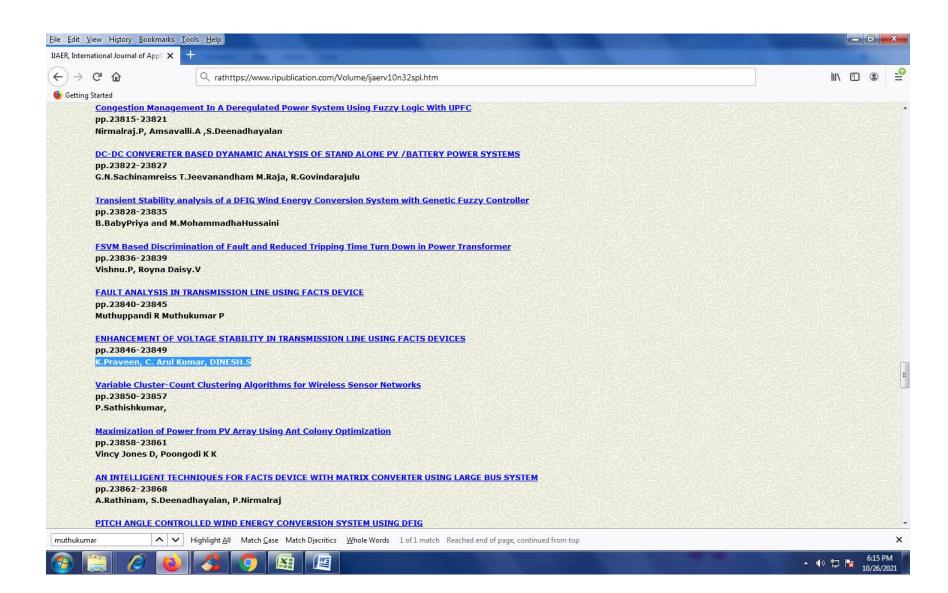




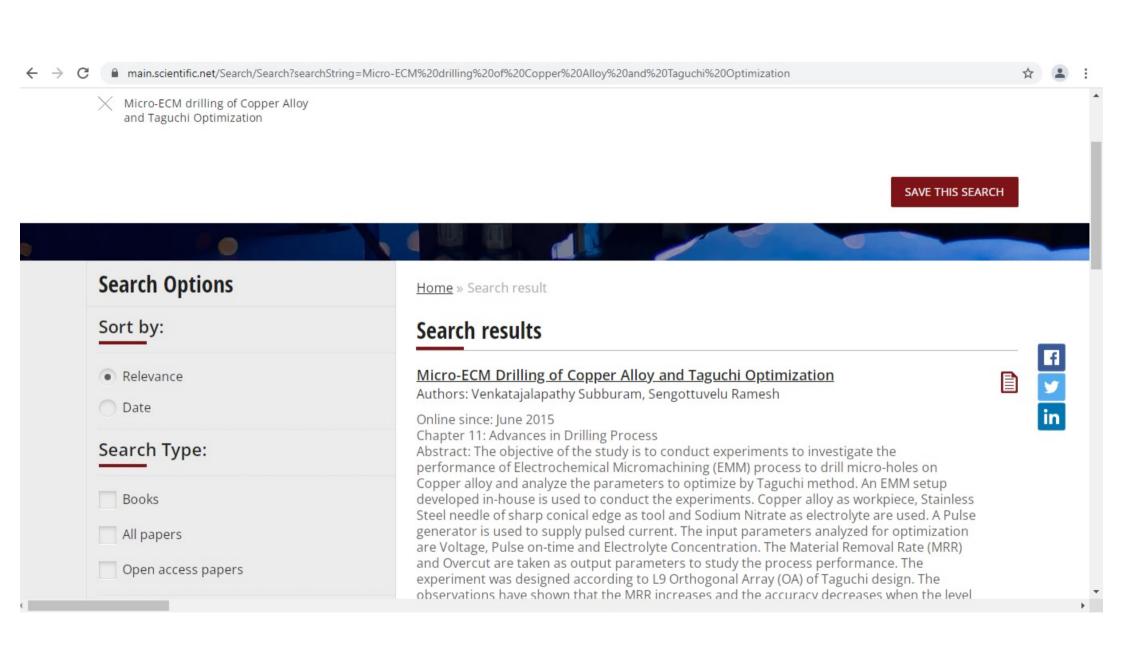


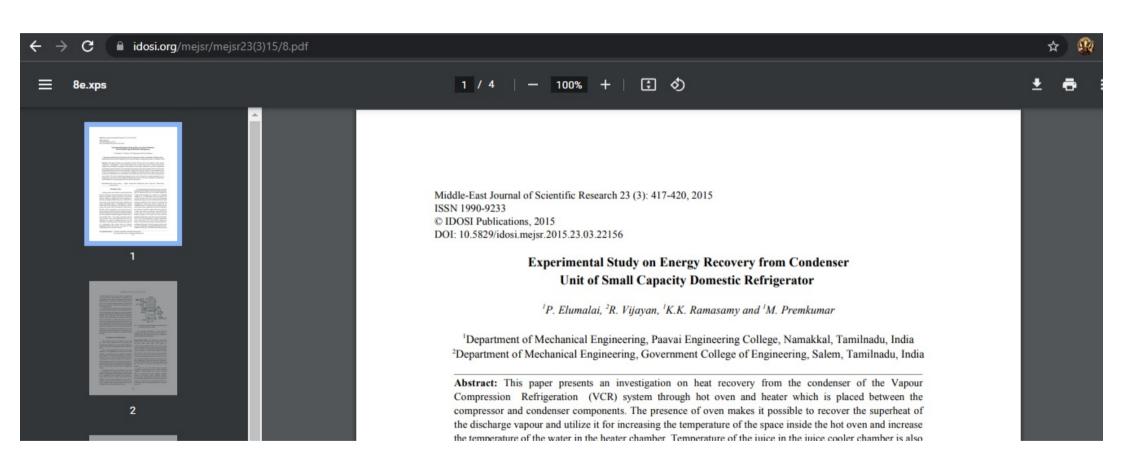


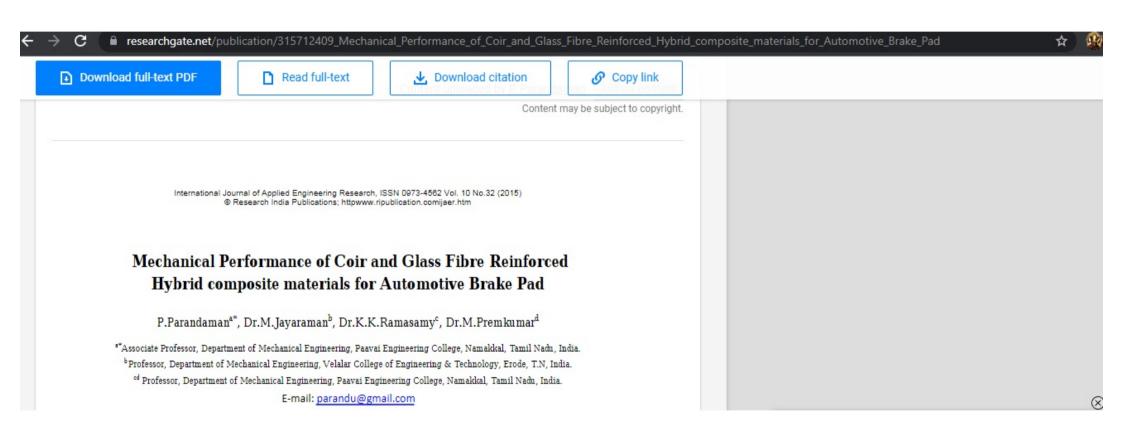


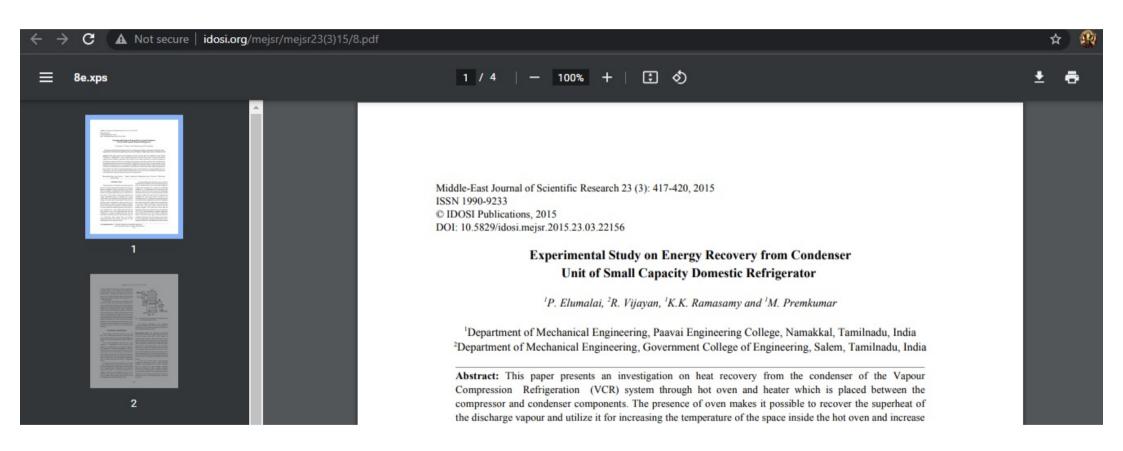


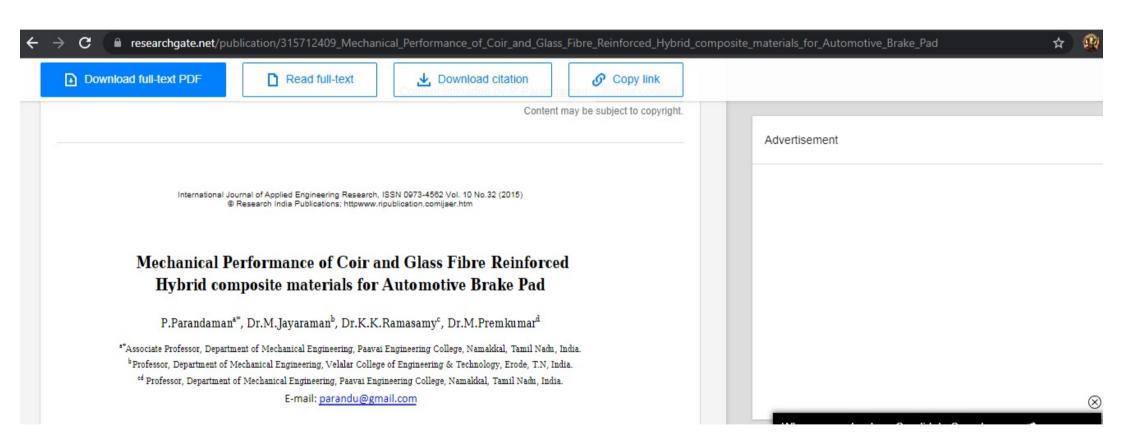


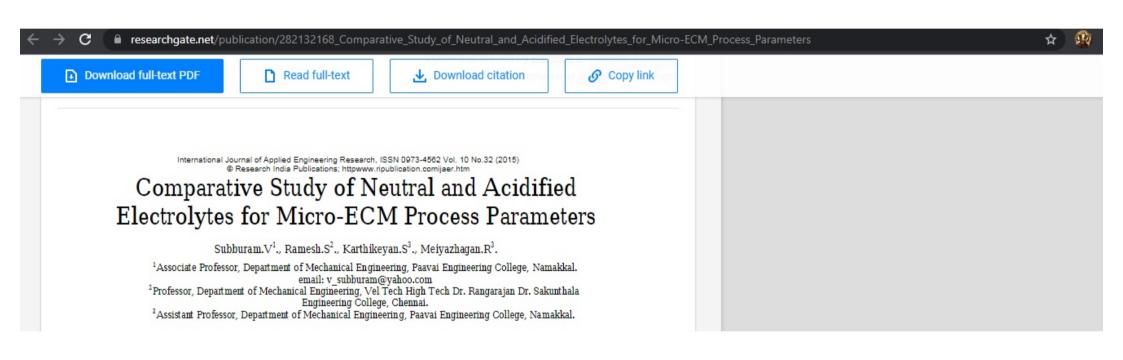


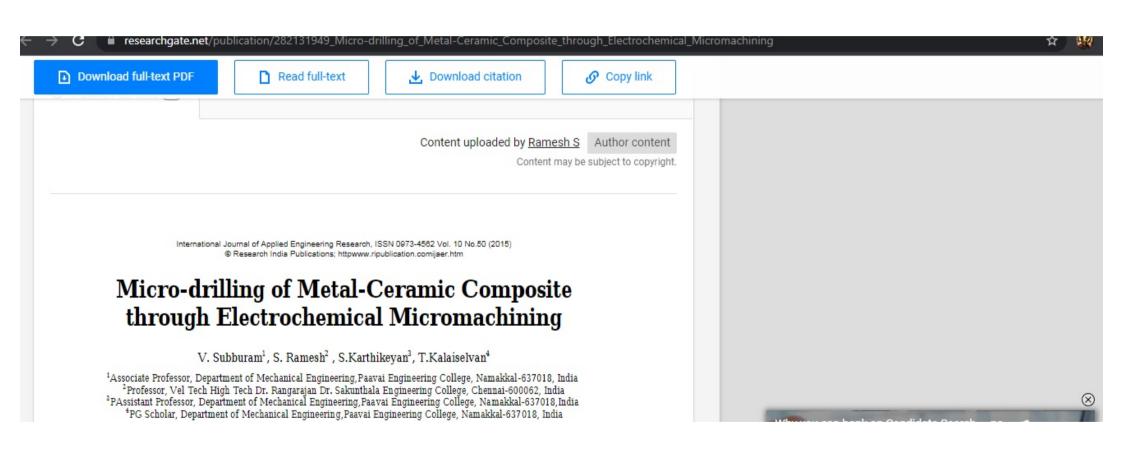


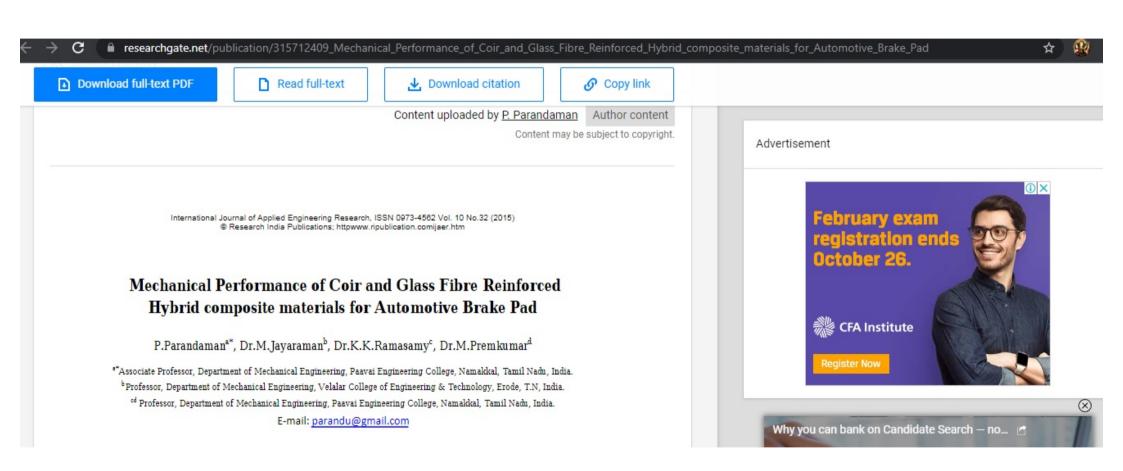


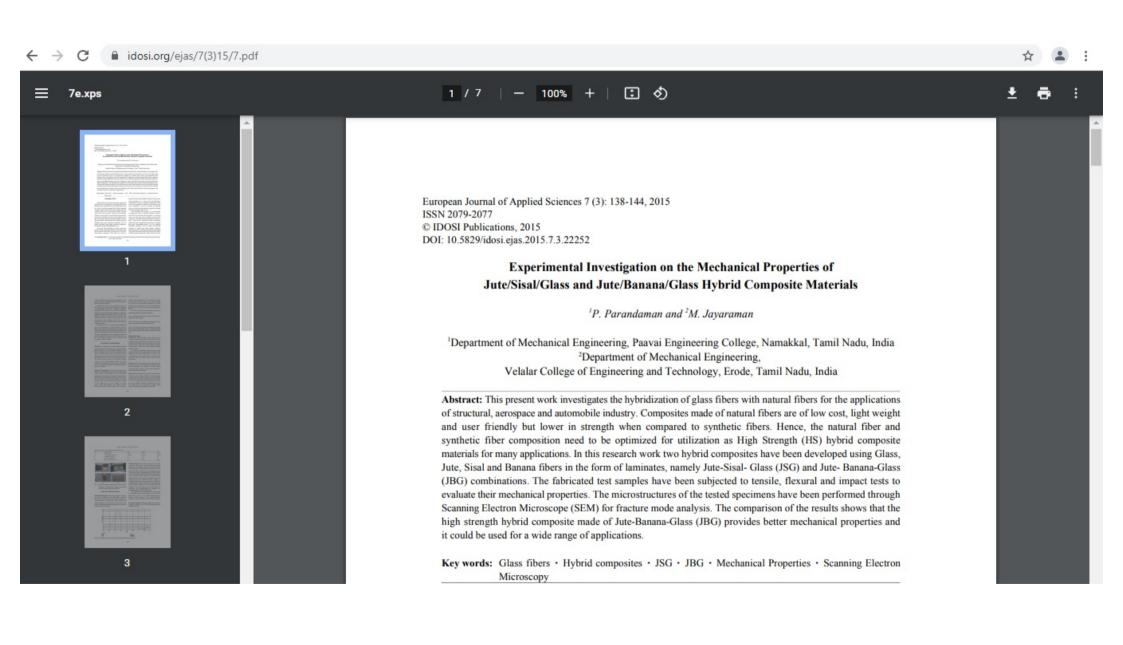


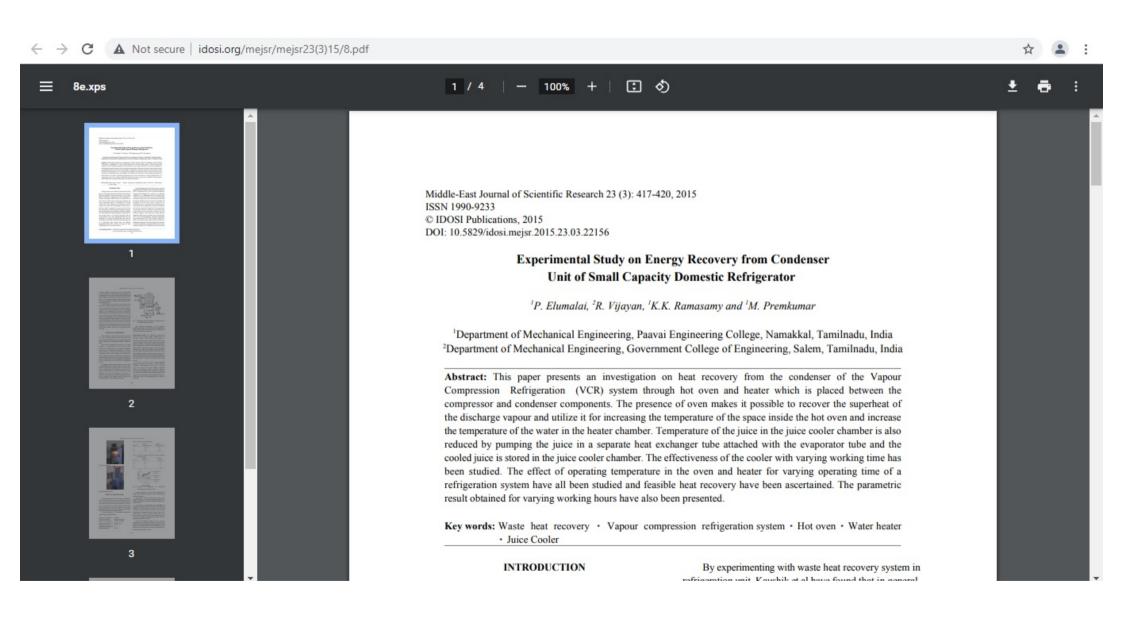






















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# Comparative Study of Neutral and Acidified Electrolytes for Micro-ECM Process Parameters

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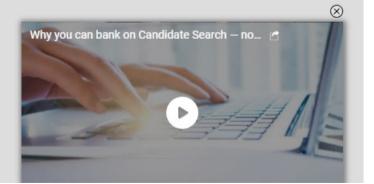
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Abstract -- Unconventional machining techniques are preferred by industries in the micro-manufacturing domain. Electrochemical Micromachining (Micro-ECM), a nonconventional method finds wide application in many industries for generating micro-features to make use of its inherent advantages. But controlling the process is a difficult task as it is a complex one with a variety of variable inputs influencing the performance. The influence of electrolyte on the performance of electrochemical micromachining process is a major factor as its characteristics like concentration, flow dynamics, temperature, throwing power and sludge formation have great influence on the process. The objective of this paper is to conduct experiments to investigate and compare the influence of neutral and acidified electrolytes on Material Removal Rate and shape accuracy. Micro-holes were generated on Stainless Steel-304 work-plate using stainless steel tool electrode. Influence of neutral salt electrolyte (NaNO2) and acidified electrolyte (NaNO2 + HCl) on process parameters like applied voltage, current, pulse ON/OFF time and electrolyte concentration are studied and compared by carrying out experiments. It is observed that acidified

Wansheng Zhao et al [4] have designed a setup along with high frequency pulse power supply system to conduct electrochemical machining experiments at micro to meso-scale. It was reported that with low machining voltage, high frequency, short pulse current, low electrolyte passivity concentration and reducing inter-electrode gap up to 10µm, better shape accuracy and material removal rate can be achieved.

Bao Huaiqian et al [5] have developed a technique called electrochemical micromachining in pure water (PW-ECM), an environment friendly micromachining for use in aerospace industry. The drawbacks in this process like short circuit and sparks were overcome by devising a combined machining process of PW-ECM assisted by ultrasonic vibration (PW-ECM/USV). Square cavities, holes and English alphabets were generated on stainless steel plate using this process.



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# Micro-drilling of Metal-Ceramic Composite through Electrochemical Micromachining

V. Subburam<sup>1</sup>, S. Ramesh<sup>2</sup>, S.Karthikeyan<sup>3</sup>, T.Kalaiselvan<sup>4</sup>

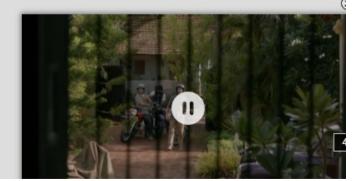
<sup>1</sup>Associate Professor, Department of Mechanical Engineering, Paavai Engineering College, Namakkal-637018, India <sup>2</sup>Professor, Vel Tech High Tech Dr. Rangarajan Dr. Sakunthala Engineering College, Chennai-600062, India <sup>3</sup>PAssistant Professor, Department of Mechanical Engineering, Paavai Engineering College, Namakkal-637018, India <sup>4</sup>PG Scholar, Department of Mechanical Engineering, Paavai Engineering College, Namakkal-637018, India

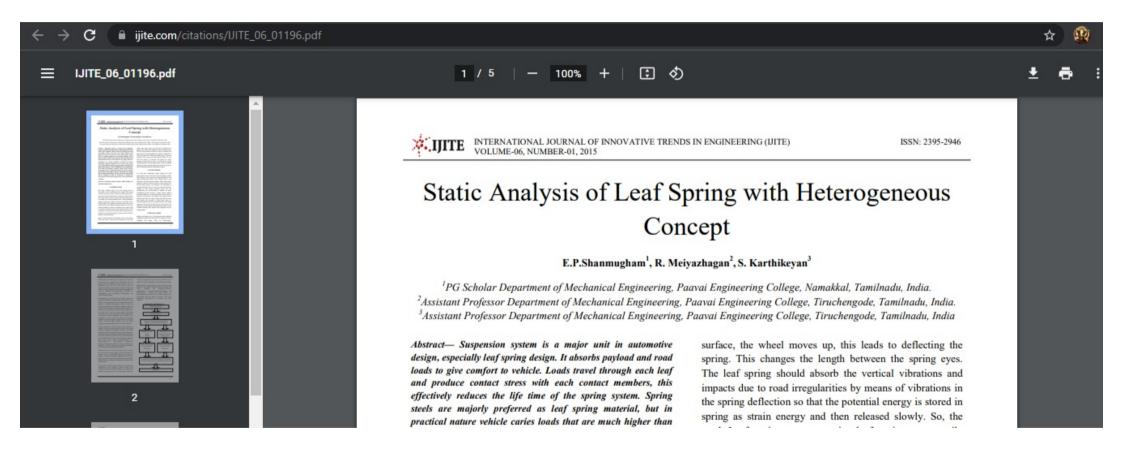
Abstract-Non-traditional machining techniques that employ other than mechanical forces are widely in use in the micromanufacturing domain to machine materials that are difficult to machine in the conventional processes. Electrochemical micromachining (EMM) is a leading unconventional technique based on the principle of electrolysis. EMM process is being researched continuously for its capability to machine a variety of conducting materials such as metals, alloys and composites. The present work involves experimental investigation of EMM process to generate micro-holes on a metal-ceramic composite specimen containing Aluminium and Titanium Carbide, A desktop EMM setup along with a pulse generator was used to conduct experiments. The influence of input parameters like machining voltage, current, pulse on-time, electrolyte concentration and frequency on Material Removal Rate (MRR) and Overcut were analysed to study the process performance. The experimental results show that presence of ceramic particles in the specimen produces poor shape accuracy.

Verwords FMM: NaNO : Al TiC composte: Material Democal

optimization of machining voltage, duty cycle and electrolyte concentration obtained from the experiments were used to attempt three dimensional micromachining. [3].

Kozak et al. investigated the detail transfer in this process from the cathodic tool electrode onto the anode work surface and studied the electrochemical copying of micro features. Application of ultra-short pulse current and ultra-small electrode gap were recommended for improving the capability of micro-ECM processes [4]. Liu Yong et al. successfully developed an EMM system, conducted experiments to find the predominant process parameters which then were applied to machine a complex microstructure [5]. Electrolytes are chosen to suit the dissolution of work material and different types of electrolytes have been tried to investigate their influence on the process. Acidified electrolytes have also been used to investigate their influence on the performance of the EMM process [6]. The absence of thermal stresses and mechanical stresses in the EMM process was successfully utilized to fabricate micro structures on NiTi Shape Memory Alloy by Joseph et al.[7]. Ramarao et al. investigated the dissolution















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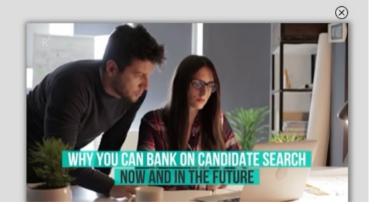
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 Professor, Vel Tech High Tech Dr. Rangarajan Dr. Sakunthala Engineering College, Chennai-600062, India
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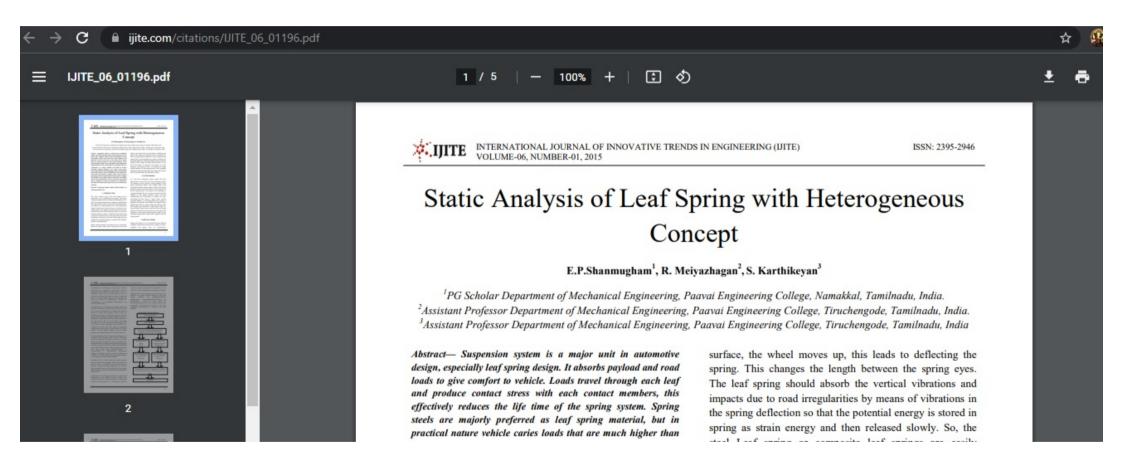
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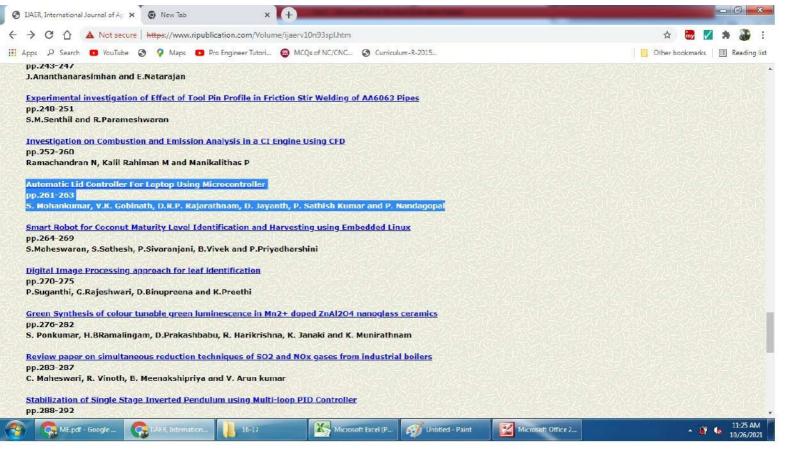
Keywords—EMM; NaNO3; Al-TiC composite; Material Removal Rate; Overcut

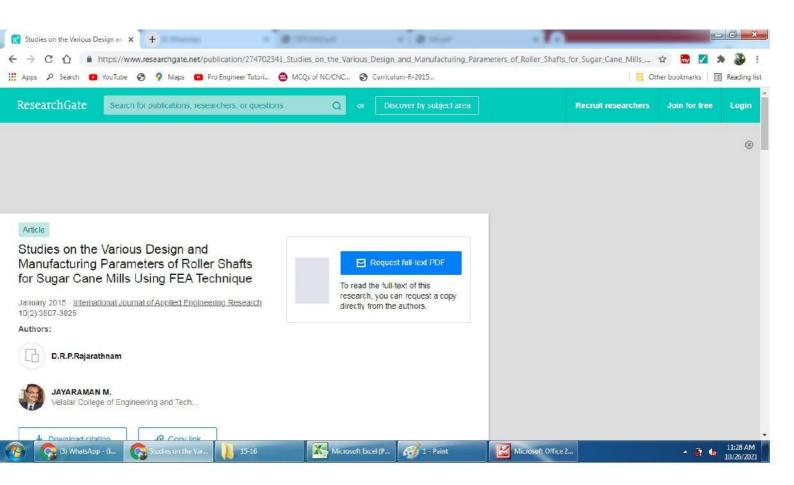
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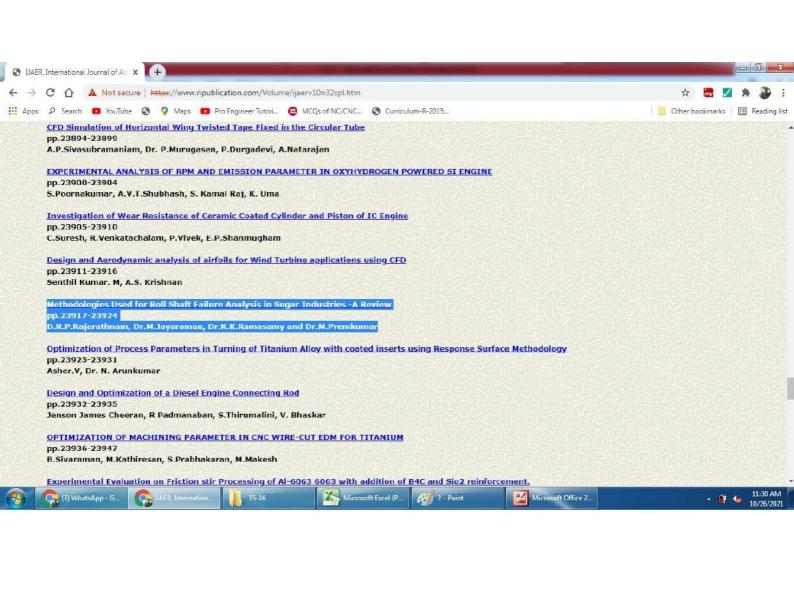
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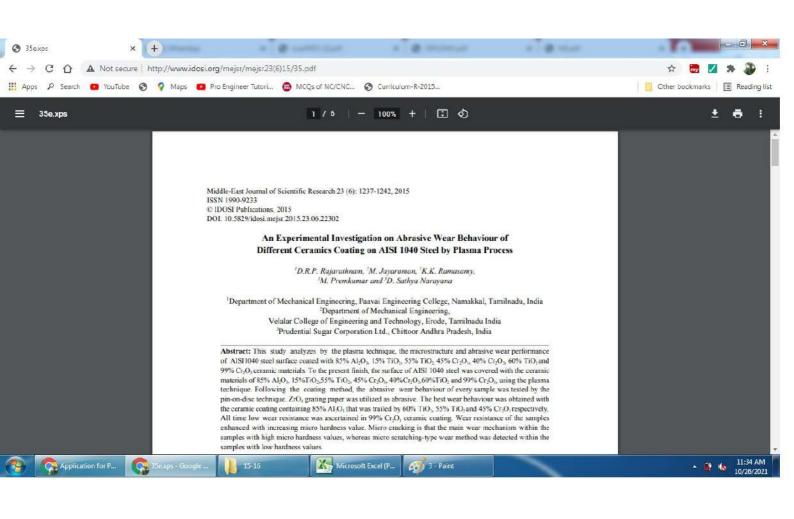


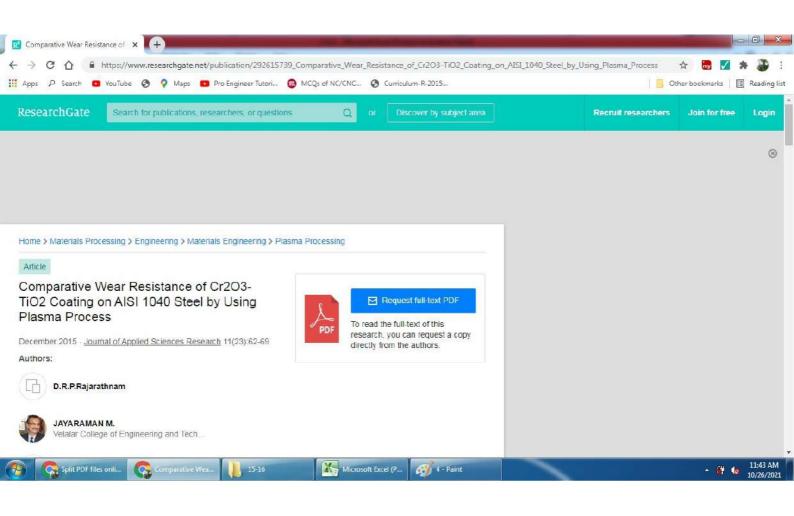


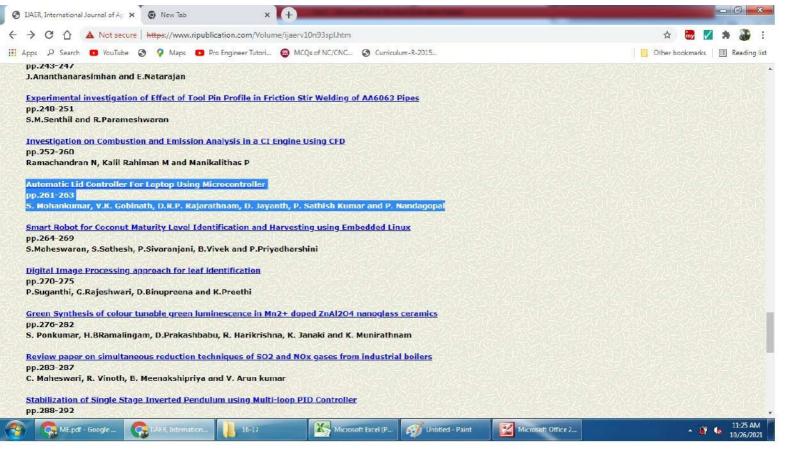






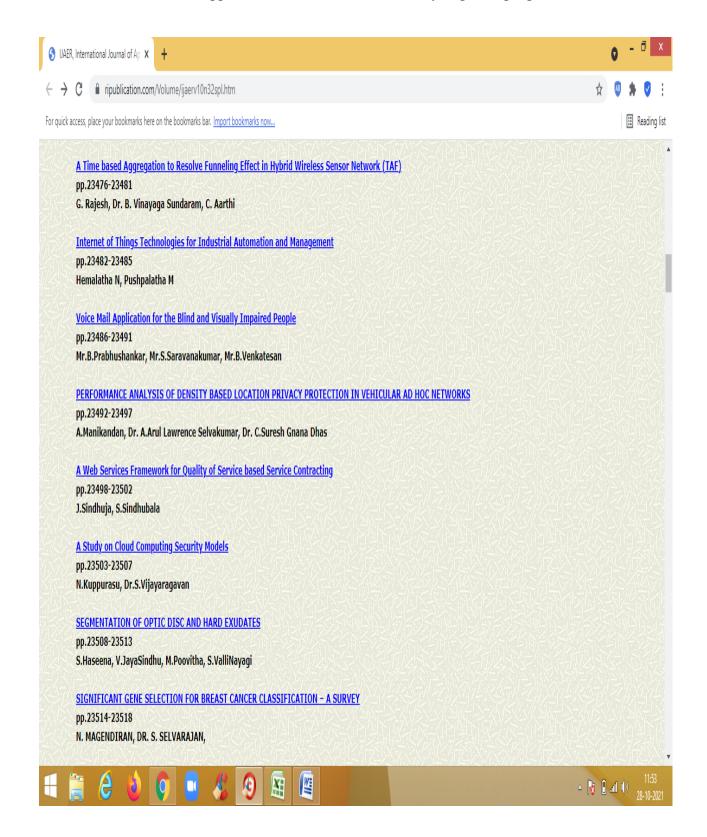




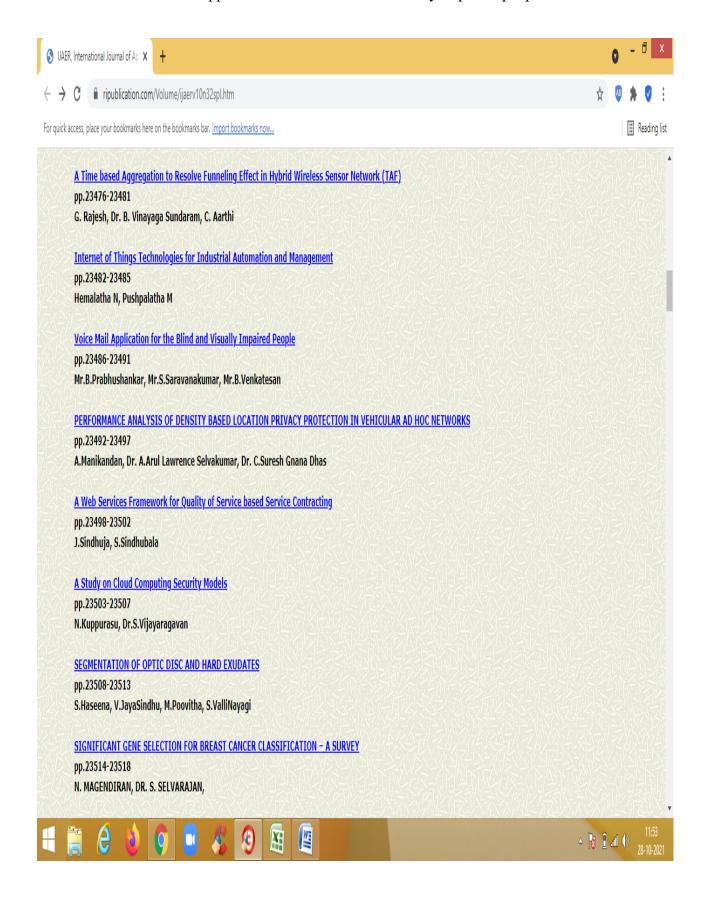


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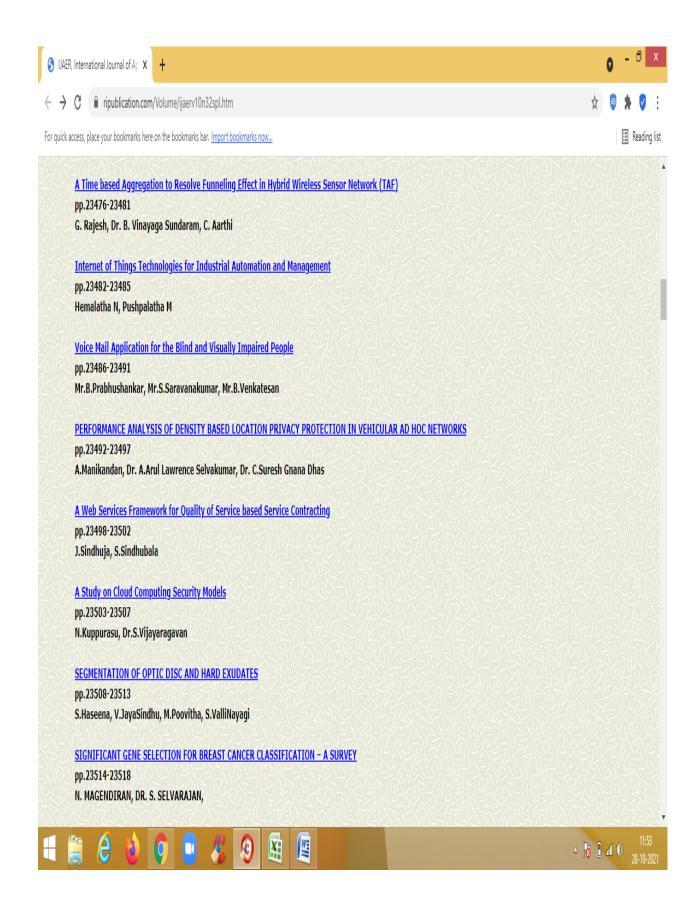
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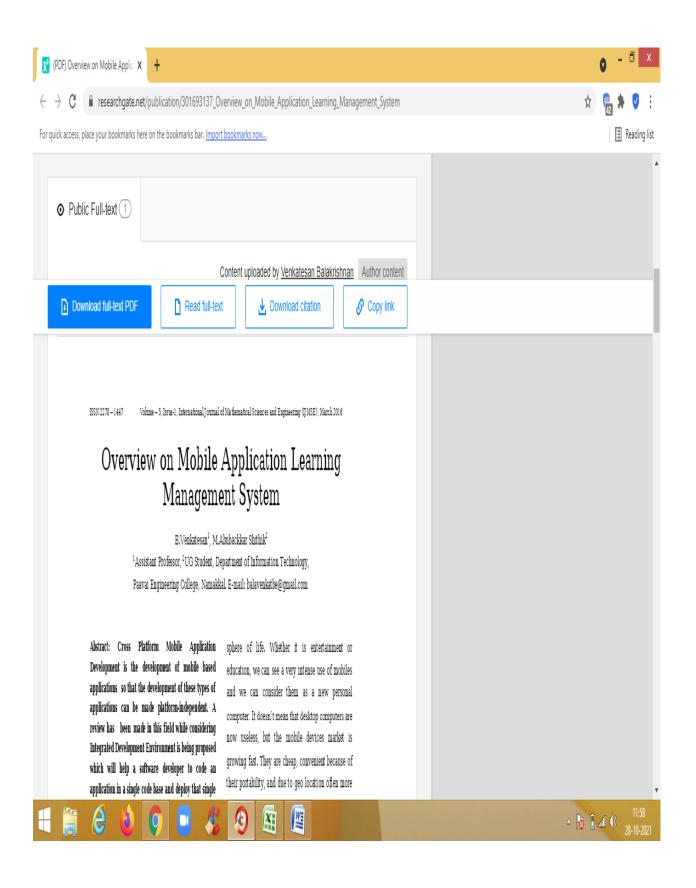
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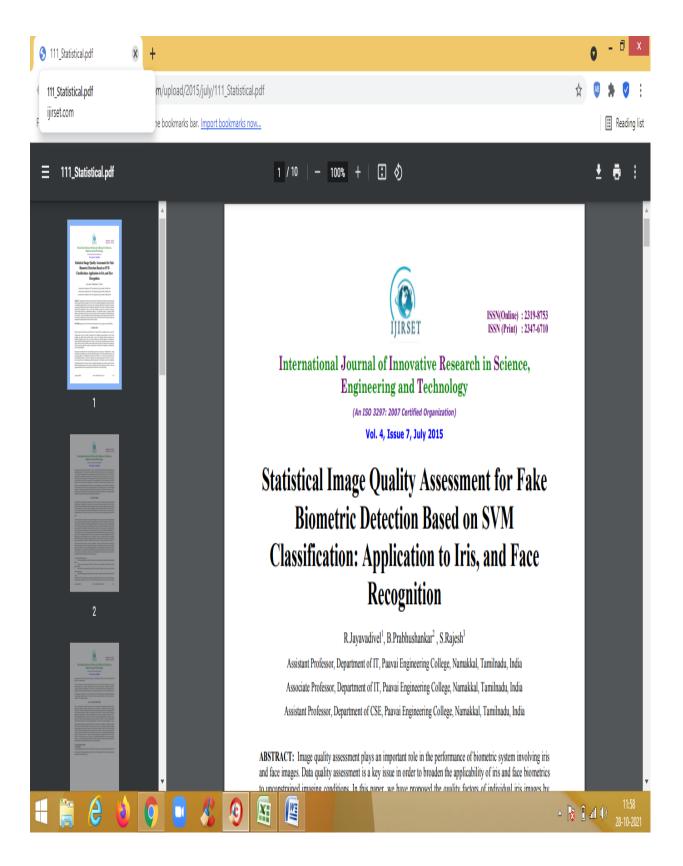
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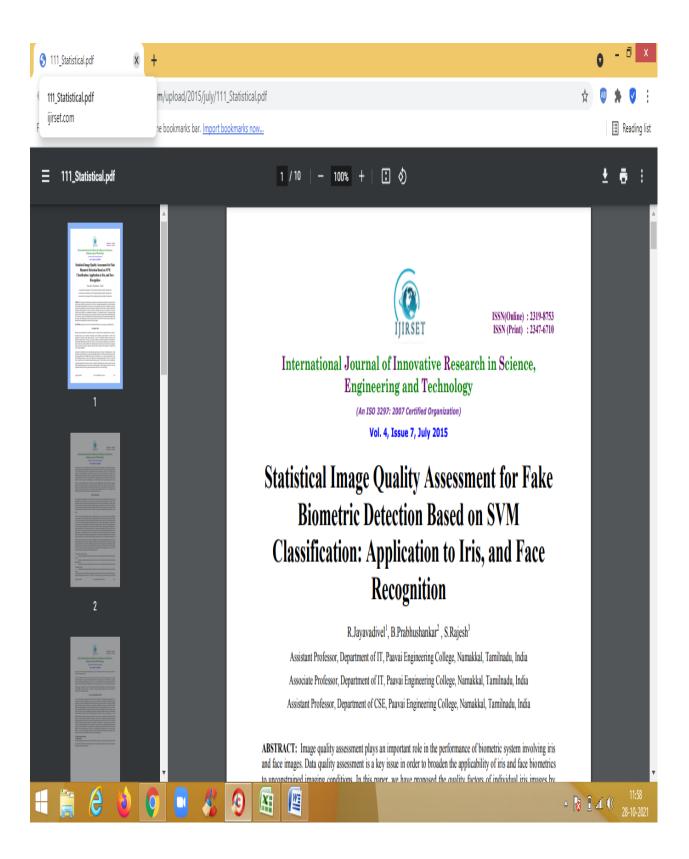
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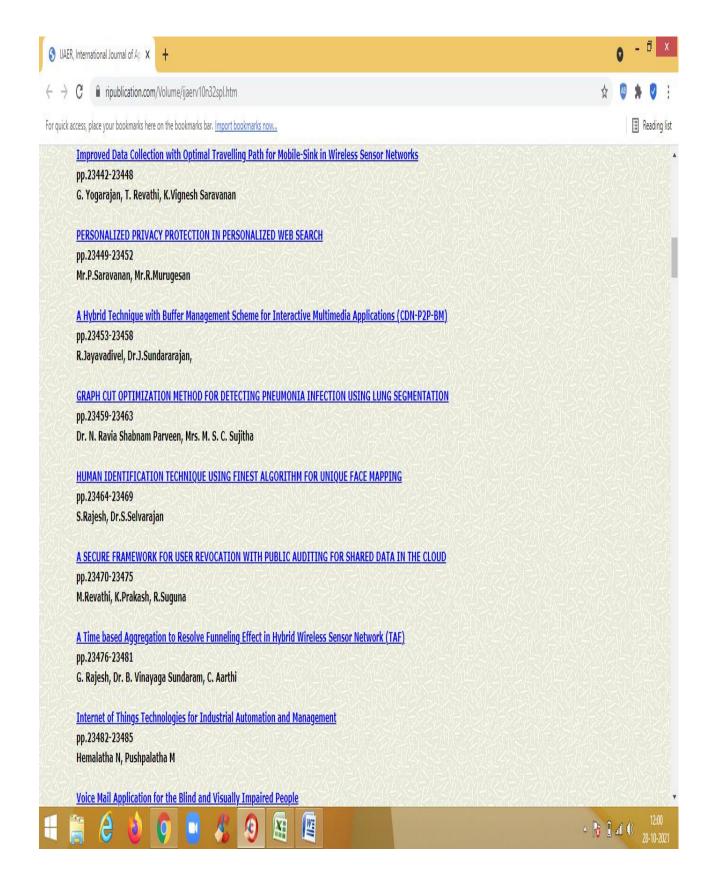
# Statistical Image Quality Assessment for Fake Biometric Detection Based on SVM Classification: Application to Iris, and Face Recognition



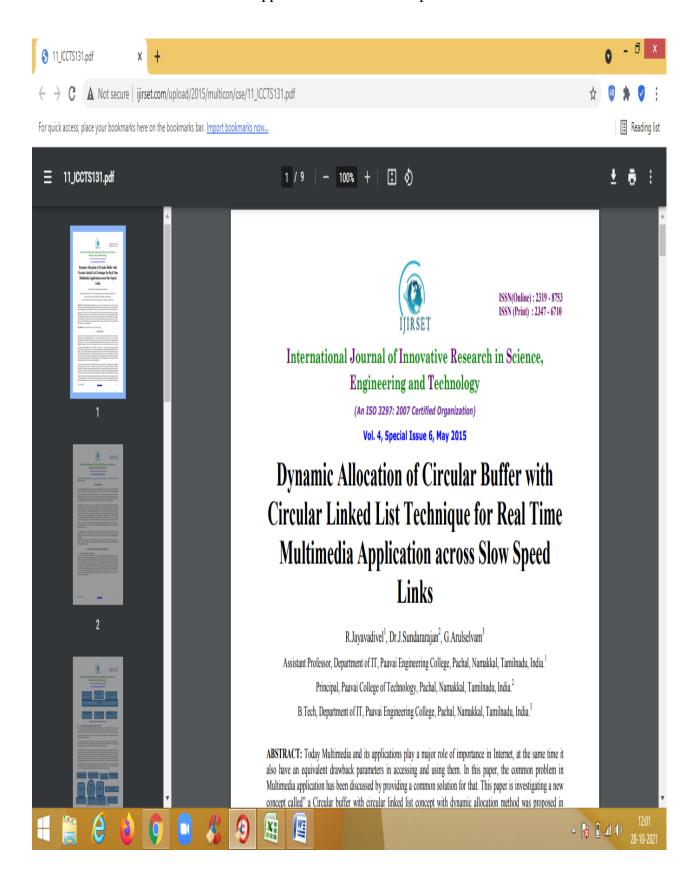
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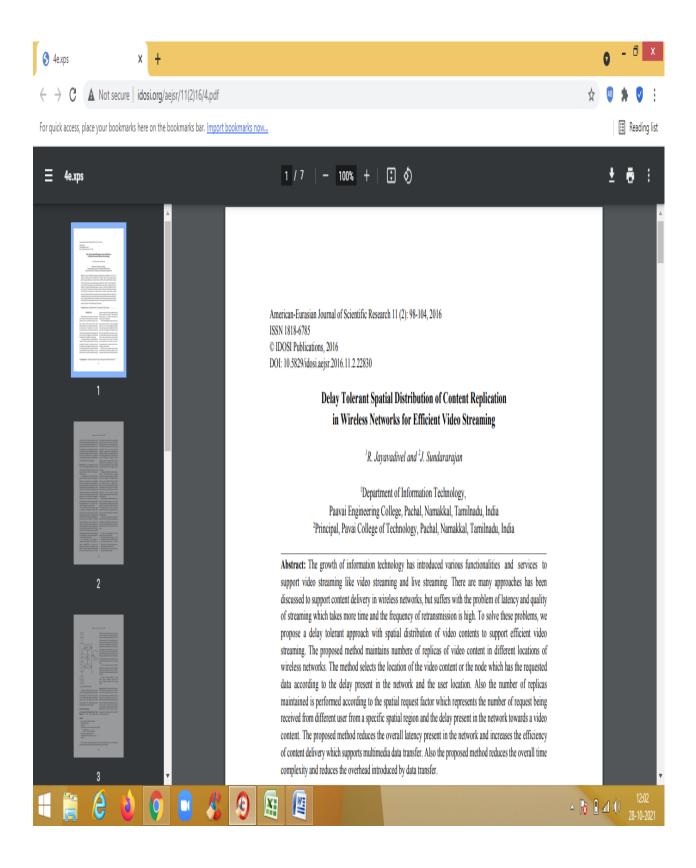
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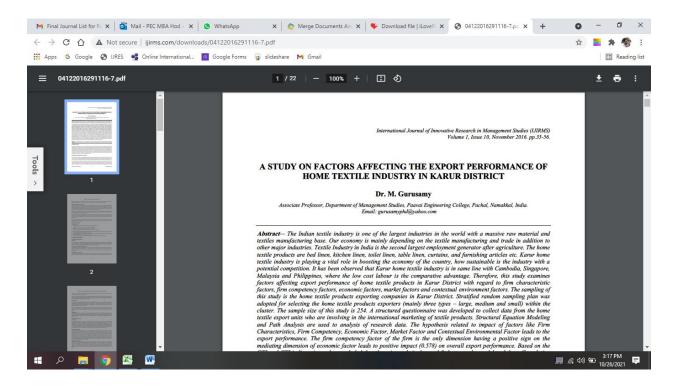


## Dynamic Allocation of Circular Buffer with Circular Linked List Technique for Real Time Multimedia Application across Slow Speed Links

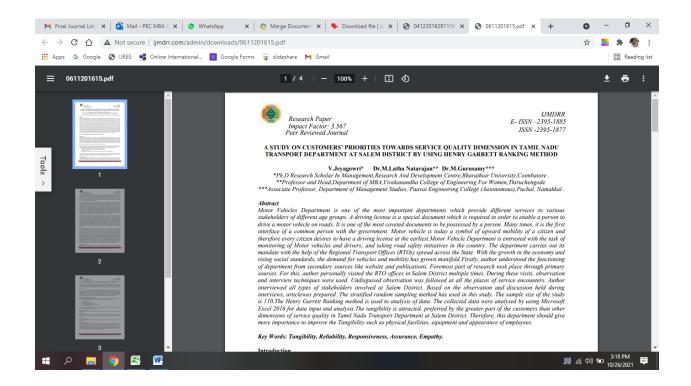


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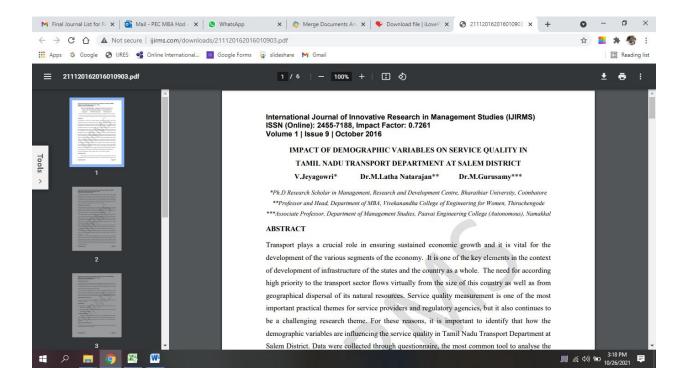




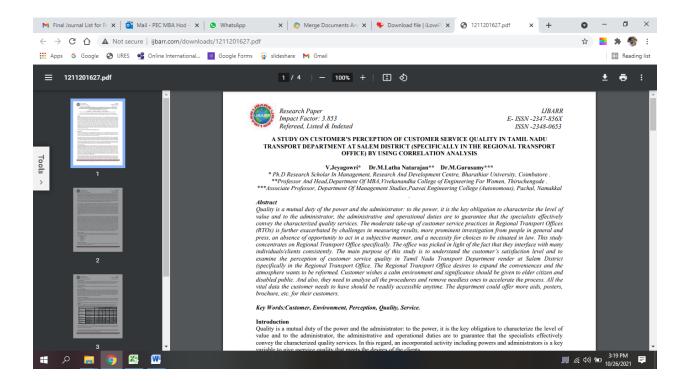
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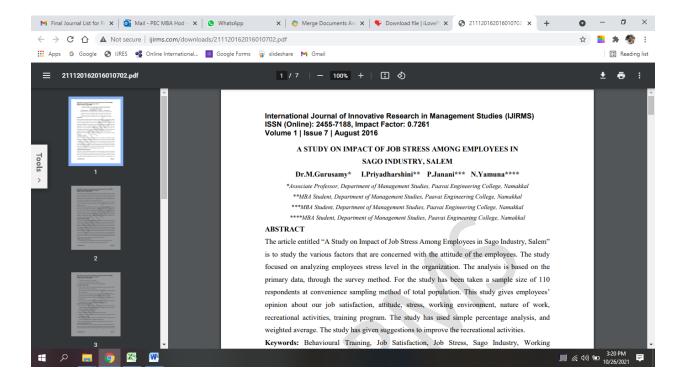
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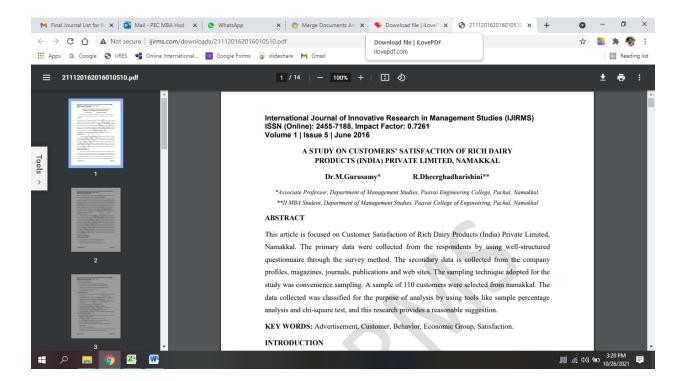
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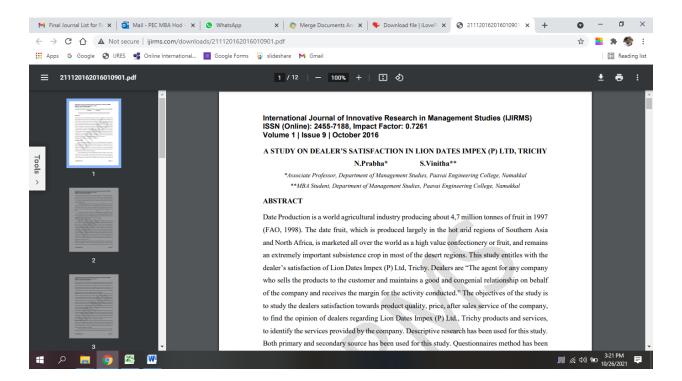
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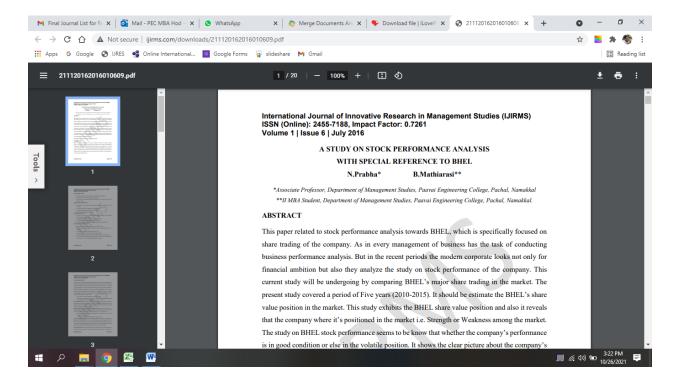
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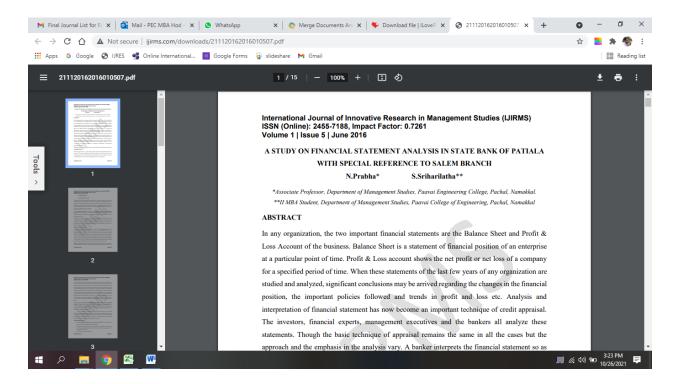
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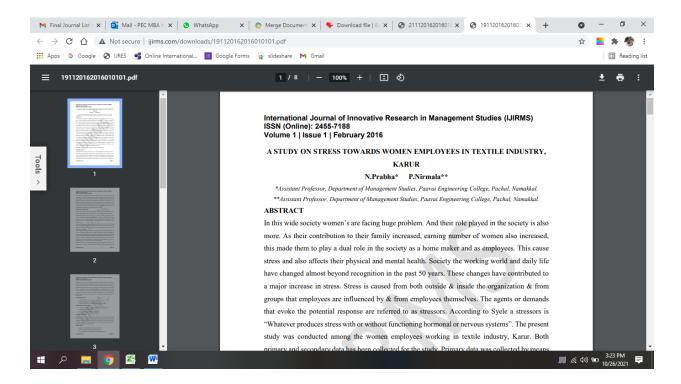
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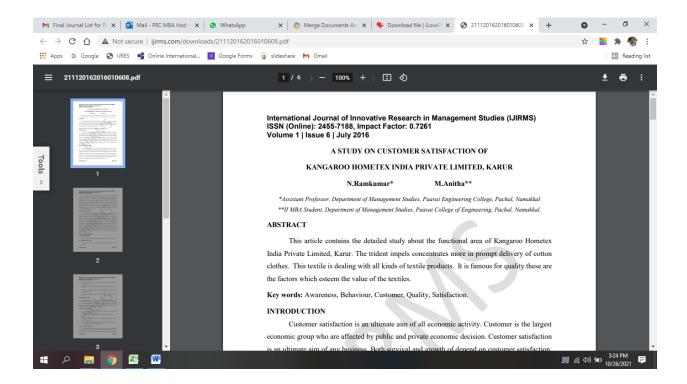
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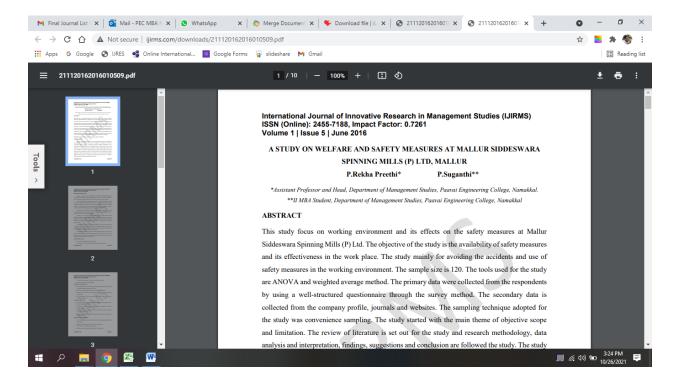
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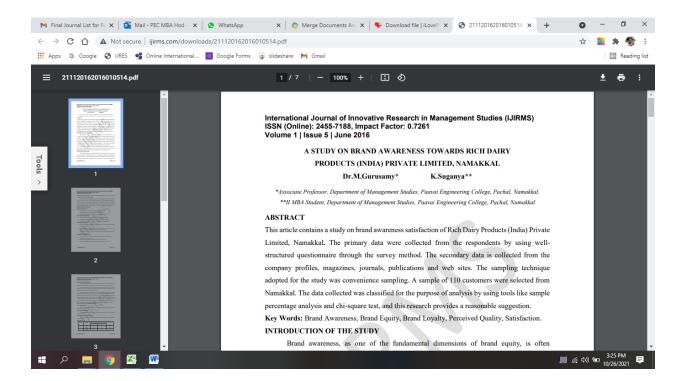
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# Effect of Mg doping on structural, optical and photocatalytic activity of SnO<sub>2</sub> nanostructure thin films

S. Vadivel · G. Rajarajan

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Abstract In the present work, magnesium (Mg) doped SnO<sub>2</sub> nanocrystalline thin films were synthesized by simple chemical bath deposition technique. The as-deposited films were annealed at 600 °C for 5 h in ambient atmosphere in order to improve crystallinity and structural perfection. The influence of Mg doping on structural, optical, and mor-

### 1 Introduction

Transparent conducting oxides (TCOs) are a unique type of materials that combine electrical conductivity and optical transparency, simultaneously, with a wide range of applications e.g. displays, low emissive (low-e) windows, thin



### Influence of Cu doping on structural, optical and photocatalytic activity of SnO2 nanostructure thin films

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Abstract This paper describes the pure and copper doped doped SuO<sub>2</sub> film was much higher than that of the pure SnO, nanocrystalline thin films with large specific surface areas fabricated on a desired substrate using a simple chemical bath deposition technique. The as-deposited films were amesled at 500 °C for 2 h in ambient atmosphere in order to improve the crystallinity and structural perfection. The influence of Cu domine on structural, ortical, and surface topography of the thin films was studied by X-ray diffraction (XRD), Raman spectra, UV-Vis spectra, photoluminescence, and atomic force micrograph images. The XRD measurements showed that films had a tetragonal rutile type structure with P42/mmw symmetry and the rebeen found to decrease with the increase of the dopont concentration as investigated by atomic force microscopy The characteristic Raman peaks observed at 325, 466, 672 The characteristic runnin peaks orserved in 32,5 408, 62, and 745 cm.<sup>22</sup> were respectively revealed infrined active (E<sub>a</sub>), Raman active (E<sub>a</sub>),  $(A_{c\phi})$  and  $(B_{c\phi})$  sibration modes of pure tetragonal rutile SnO<sub>2</sub> structure. The optical band gap energy of pure SnO<sub>3</sub> has been found to be in the range of 3.68 eV and it is shifted to 3.32 eV for 10 wt% Cu doning. The photocatalytic activities of the films were evaluated by the degradation of methylene blue rhodumine B in an aqueous solution under ultraviolet light irradiation. The photocatalytic activity and reusability of Cu (10 we%)

Textiles industry wastewater is heavily charged with unconsumed dyes, surfactants and sometimes traces of metals. These offluents cause a lot of damage to the environment. In most countries researchers are looking for appropriate treatments in order to remove pollutants, impurities and to obtain the decolourization of dye house effluents [1]. Metal sulfs were good in agreement with the standard JCPDS data (cast no. 41-1445). The surface roughness and porosity has
show lower light-harvesting ability in visible light [2]. show lower light-harvesting ability in visible light [2]. Therefore, coupling of semiconductors with different band gaps is a good approach to prepare photo catalysts with high activity and good stability. Among the large number of metal oxides (TiO2, ZnO, WO3 and In2O3), tin oxide (SnO2) has become a promising material due to its unique proporties such as high electrical conductivity, high optical trans-parency in the visible part of the electromagnetic spectrum. This metal oxide has wide range of applications in low emission glass, electrodes, organic light emitting diodes optoelectronic devices, lithium batteries, gas sensors, heat reflectors and polymer based electronics [3]. Moreover, SnO<sub>2</sub> exhibits good activity and stability under irradiation. However, the pure SnO2 shows much lower photo catalytic activity even under UV irradiation due to its large band gap (3.6 eV) [4]. To improve its photo catalytic activity, it is necessary to couple SuO, with another semiconductor with lower band gap. The CuO is a p-type semiconductor with a small band gap (1.7-1.2 eV). If the SnO<sub>2</sub> is coupled with the CuO, the n-SnO<sub>2</sub>/p-CuO beterojunctions can be formed in. The photo generated electrons from SnO<sub>2</sub> can easily migrate to CuO. This favors the separation of photo generated

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### Effect of W doping on structural, optical and photocatalytic activity of SnO2 nanostructure thin films

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Abstract Thin films of undoped and tungsten (W) doped 1 Introduction SnO, have been synthesized on a glass (ITO) substrate using a simple chemical bath deposition technique. The as-deposited films were amealed at 500 °C for 5 h in ambient atmosphere in order to improve the structural perfections and crystallinity. X-ray diffraction studies showed that the W: SnO<sub>2</sub> films were polycrystalline in nature with tetragonal ratile type structure of SnO2 phase. The surface roughness and porosity have been found to decrease from 47 to 39 nm with the increase of dopart concentration as investigated by atomic force microscopy. The optical absorption edge was found to be 3.46 eV, while the higher concentration of W doped films shifted towards lower energy (red shift) in the same of 3.35 eV. The effect of W on impurity, defect states and oxygen vacancy of the SeO2 film was analyzed by photoluminescence spectra. The photocatalytic activities of the films were evaluated by the degradation of methylene blue (MB) rhodamine B (RHB) in an aqueous solution under visible light irradiation. The photocatalytic activity and reusability of W (10 wt%) doped SnO<sub>2</sub> films were much higher than that of the pure SnO<sub>2</sub>. The improvement mechanism of RHB by W-SnO<sub>2</sub> catalyst was also discussed.

In recent years, transparent conducting exide (TCO) films like, SnO<sub>2</sub>, ZnO, TsO<sub>2</sub> and WO<sub>3</sub> have reached a vital place in a variety of optoelectronic devices such as solar cell, gas sensor, flat panel displays, and varistors [1]. Hence, there has been a growing attention in the applications of TCO films in solar cell device. Among these, SnO2 serves as an important material due to its excellent chemical stability. optical and electrical properties.  $SnO_2$  is an important n-type semiconductor with a wide band gap (Eg = 3.6 eV, at 27 °C) and it is well known for various potential applications such as excellent gas sensors, electrode materials in Li/SnO: butteries, catalysts [2] and so on. In addition to this, they exhibit low electrical resistivity and high electrical transmittance in the visible light and near IR region. Many methods have been adopted to synthesize of SnO2 thin films such as, sol-gel [3], pulsed plasma deposition [4], pulsed laser deposition [5], reactive evaporation [6] and chemical bath deposition [7] methods. Among these techniques, chemical bath deposition method is an attractive method to get intended thin films. Moreover chemical bath deposition yields stable, adherent, uniform and hard films with good reproducibility by a relatively simple process [7]. So in the present work, we adopted chemical both deposition method to synthesize of pure and W doped SnO<sub>2</sub> thin films.

Generally, SnO<sub>2</sub> suffers from low photocatalytic efficiency because of its wide-hand gap (energy of the hand gap is about 3.6 eV) and high recombination rates of photo generated electron-hole pairs. This deficiency binders SnO<sub>2</sub> photocatalyst using widely and practically in the environmental application [8]. To overcome this moblem doped with suitable metal ions can increase the photocatalytic activity. Many researchers have paid much attention

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